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Key indicators

Single-crystal X-ray study T = 293 K Mean σ (C–C) = 0.005 Å R factor = 0.029 wR factor = 0.086 Data-to-parameter ratio = 8.0

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e. Diphenyl(1,1,3,3-tetramethyl-2,3-dihydro-1*H*-isoindol-2-yl)phosphine oxide

The title compound, $C_{24}H_{26}NOP$, crystallizes with two independent non-chiral molecules in the asymmetric unit. The molecular structure consists of a tetrahedral P atom bonded by two phenyl groups, a phosphate O atom and an isoindole group. The geometry of the two molecules is similar but not identical.

Comment

The title compound, (I), was obtained as part of an investigation of the use of the radical scavenger 1,1,3,3-tetramethyl-2,3-dihydro-1*H*-isoindol-2-yloxyl to study the reaction of phosphorus-centred radicals with alkenes and alkynes (Busfield *et al.*, 1994; Bottle *et al.*, 1994).



Experimental

The title compound was obtained in 6.3% yield from the reaction of diphenylphosphine (100 mg), di-*tert*-butyl peroxyoxalate (DTBP) (58 mg) and benzene (2.5 ml) (solvent) in the presence of the radical scavenger (103 mg). Reaction mixtures were degassed using repeated freezing/evacuating/thawing cycles on a high vacuum line, then sealed under vacuum in glass and heated for ten half-lives of the initiator, DTBP (68 min at 330 K). Interestingly, in the identical reaction to that above but with no initiator (DTBP) present, the title compound was still obtained (21%), along with the same additional five phosphorus-containing compounds. A small quantity of the title compound was taken up in methanol (HPLC grade) to produce a concentrated sample. One small drop of water (HPLC grade) was then added to the solution. The solution was allowed to stand in the freezer at 253 K (length of time unrecorded) to produce X-ray quality crystals.

Crystal data
C24H26NOP
$M_r = 375.43$
Monoclinic, P21
a = 14.48 (2) Å
b = 8.323(1) Å
c = 18.455 (5) Å
$\beta = 112.280 \ (12)^{\circ}$
$V = 2058 (3) \text{ Å}^3$

Z = 4 $D_x = 1.212 \text{ Mg m}^{-3}$ Mo K α radiation $\mu = 0.15 \text{ mm}^{-1}$ T = 293 (2) K Prism, colourless $0.35 \times 0.22 \times 0.13 \text{ mm}$ Received 29 April 2006 Accepted 9 May 2006

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Figure 1

The molecular structure of molecule A, showing displacement ellipsoids drawn at the 50% probability level. H atoms are shown as small spheres

3878 independent reflections

3276 reflections with $I > 2\sigma(I)$

 $R_{\rm int} = 0.017$

 $\theta_{\rm max} = 25.0^\circ$

3 standard reflections

frequency: 120 min

intensity decay: 2%

Data collection

Enraf-Nonius CAD-4 diffractometer ω -2 θ scans Absorption correction: ψ scan (North *et al.*, 1968) $T_{\min} = 0.950, T_{\max} = 0.981$ 4422 measured reflections

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_0^2) + (0.0591P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.029$	+ 0.1682P]
$wR(F^2) = 0.086$	where $P = (F_0^2 + 2F_c^2)/3$
S = 1.03	$(\Delta/\sigma)_{\rm max} < 0.001$
3878 reflections	$\Delta \rho_{\rm max} = 0.18 \ {\rm e} \ {\rm \AA}^{-3}$
487 parameters	$\Delta \rho_{\rm min} = -0.18 \text{ e} \text{ Å}^{-3}$
H-atom parameters constrained	Absolute structure: Flack (1983), 0
	Friedel pairs
	Flack parameter: 0.03 (8)

H atoms were placed in calculated postions, with C–H = 0.93 (aromatic) or 0.96 Å (methyl), and included in the refinement in riding-model approximation, with $U_{\rm iso}(\rm H) = 1.2 U_{eq}(\rm C)$, or $1.5 U_{eq}(\rm C)$ for methyl H atoms.

Data collection: *SDP* (Frenz, 1985); cell refinement: *SDP*; data reduction: *WinGX* (Farrugia, 1999); program(s) used to solve struc-



Figure 2

The molecular structure of molecule B, showing displacement ellipsoids drawn at the 50% probability level. H atoms are shown as small spheres

ture: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97*; molecular graphics: *PLATON* (Spek, 2003); software used to prepare material for publication: *SHELXL97*.

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References

- Bottle, S., Busfield, W. K., Grice, I. D., Heiland, K., Jenkins, I. D., Meutermans, W. & Monteiro, M. (1994). Prog. Pacific Polym. Sci. 48, 85–97.
- Busfield, W. K., Grice, I. D. & Jenkins, I. D. (1994). Aust. J. Chem. 48, 625–634. Farrugia, L. J. (1999). J. Appl. Cryst. 32, 837–838.
- Flack, H. D. (1983). Acta Cryst. A**39**, 876–881.
- Frenz, B. A. (1985). Enraf-Nonius SDP Structure Determination Package. Version 3.0. Enraf-Nonius, Delft, The Netherlands.
- North, A. C. T., Phillips, D. C. & Mathews, F. S. (1968). Acta Cryst. A24, 351–359.
- Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
- Spek, A. L. (2003). J. Appl. Cryst. 36, 7-13.